ANIONIC SURFACTANTS AS MBAS SM 20 th Ed. 5540 C						
Facility Name:	VELAP ID					
Assessor Name:Analyst Name:	Inspection Date					
Relevant Aspect of Standards	Method Reference	Y	1 N	Ά	Comments	
Records Examined: SOP Number/ Revision/ Date	Analyst:					
Sample ID: Date of Sample Prepare	ration:	Date of Analysis:				
Were stock LAS solutions made so that 1 mL of them contained 1 mg LAS and prepared weekly?	3.a					
Were standard LAS solutions made so the 1 mL of them contained 10 µg LAS and prepared daily?	3.b					
Were methylene blue reagents composed of 100 mg methylene blue and 41 mL 6N $\rm H_2SO_4$ and 50 g $\rm NaH_2PO_4$ • $\rm H_2O$ in 1000 mL $\rm H_2O$?	3.g					
Were wash solutions composed of 41 mL 6N H_2SO_4 and 50 g NaH_2PO_4 • H_2O in 1000 mL?	3.h					
Was reagent grade water used MBAS-free?	4.i					
Were calibration curves composed of at least 5 standards?	4.a					
Did calibration curves have correlation coefficients of 0.995 or better?	4.a					
Were check standards at the reporting limit analyzed daily?	4.a					
Were check standards above expected sample concentration analyzed daily?	4.a					
Did check standards at the reporting limit fall within $\pm 25\%$ of original values?	4.a					
Did other check standards fall within ±10% of original value?	4.a					
Were samples where interferences were expected sublated by extracting with methanol and nitrogen gas and then heating to dryness prior to the re-addition of water?	4.b					
Were samples made alkaline after placing in separatory funnels by the addition of 1N NaOH with the use of phenolphthalein indicator?	4.b					
Notes/Comments:		. 1	•			

ANIONIC SURFACTANTS BY MBAS SM 20^{th} Ed. 5540 C

Relevant Aspect of Standards	Method Reference	Y	N	N/A	Comments
Were a few drops of 30% H ₂ O ₂ added to samples where sulfides decolorized methylene blue?	4.c				
Did samples then have their pink color discharged by the addition of 1N H_2SO_4 ?	4.d.1				
If at any time after the addition of 25 mL of methlyene blue reagent any samples lost their blue color during extraction, were such samples discarded, and extraction repeated with smaller volumes?	4.d.2 4.d.3				
If consistent emulsions formed during extractions with CHCl ₃ , were those emulsions broken by addition of <10 mL of isopropyl alcohol, and then was that same volume of isopropyl alcohol added to all standards?	4.d.2				
Were samples extracted three times with CHCl ₃ ?	4.d.3				
Were all three CHCl ₃ extracts from each sample combined into a separatory funnel, shaken for 30 seconds with 50 mL of wash solution, and allowed to settle?	4.d.4				
Were CHCl ₃ layers then drawn off from the separatory funnels through plugs of glass wool, and the wash solutions extracted twice with CHCl ₃ with the CHCl ₃ layers being drawn through the glass wool into the same vessel?	4.d.4				
Were aborbances of extracts determined at 652 nm against a blank of CHCl ₃ ?	4.e				
Were calculations made correctly?	5				

Notes/Comments: